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solution slowly
until some Na₂
SO₄ · 10H₂O
crystallizes
out. Calculate

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the g of Na_2

$\text{SO}_4 \cdot 10\text{H}_2\text{O}$

recovered in the
crystals per 100
g of initial
solution, if the
residual.

solution after
the crystals are
removed contains
28% Na_2SO_4 .

Right answer

but: -10 if

answer is in g

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of Na_2SO_4 and
not g of $\text{Na}_2\text{SO}_4 \cdot 10\text{H}_2\text{O}$

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F.1 Heats of
Formation and
Heats of
Combustion of
Compounds at
250C*t Standard

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states of
products for are
 $\text{CO}_2(\text{g})$, $\text{H}_2\text{O}(\text{l})$,
 $\text{N}_2(\text{g})$, $\text{SO}_2(\text{g})$,
and $\text{HCl}(\text{aq})$.

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calculations are
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tracing the
inflow and
outflow of
material in a
process, and
thus establish
the quantities
of various
materials in
each process
stream. The
procedure is...

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