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1. On the SIMULATION pane, select Copper.

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Calorimetry Lab Answers Correct Answer: A. Substance A A chemist mixes 500 g of lead at 500°C with 1,200 g of water at 20°C. She then mixes 500 g of copper at

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500°C with 1,200 g of water at 20°C. The specific heat capacity of lead is 0.1276 J/g°C and the specific heat capacity of copper is 0.3845 J/g°C.

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Investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water. Modify initial mass and temperature values to see effects on the system. One or any combination of the substances can be mixed with water. A dynamic graph (temperature vs. time) shows temperatures of the individual substances after mixing.

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Gizmo Warm-up A calorimeter is

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an insulated container filled with a liquid, usually water. When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water heats up. Calorimeters can be used to find a substance's specific heat capacity.

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Student Exploration Calorimetry Lab Gizmo Answer Key

Calorimeters can be used to find a substance's specific heat capacity. You will use the Calorimetry Lab Gizmo to determine the specific heat capacities of various substances.

1. On the SIMULATION pane, select Copper.

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